

<b>Component</b>	<b>Formula</b>	<b>Molar %</b>
methane _____	CH <sub>4</sub>	90
ethane _____	C <sub>2</sub> H <sub>6</sub>	5
propane _____	C <sub>3</sub> H <sub>8</sub>	4
n-butane _____	C <sub>4</sub> H <sub>10</sub>	1
isobutane [2-methylpropane] _____	C <sub>4</sub> H <sub>10</sub>	
n-pentane _____	C <sub>5</sub> H <sub>12</sub>	
isopentane [2-methylbutane] _____	C <sub>5</sub> H <sub>12</sub>	
n-hexane _____	C <sub>6</sub> H <sub>14</sub>	
n-heptane _____	C <sub>7</sub> H <sub>16</sub>	
ethylene [ethene] _____	C <sub>2</sub> H <sub>4</sub>	
propylene [propene] _____	C <sub>3</sub> H <sub>6</sub>	
1-butene [n-butylene] _____	C <sub>4</sub> H <sub>8</sub>	
hydrogen sulfide [hydrogen sulphide]	H <sub>2</sub> S	
nitrogen _____	N <sub>2</sub>	
oxygen _____	O <sub>2</sub>	
carbon dioxide _____	CO <sub>2</sub>	
<b>LNG Lab Calc</b>	<b>Total, Molar %</b>	<b>100</b>

Molar Mass	18.287
Ideal Gas Gross Calorific Value by volume, MJ/m <sup>3</sup> at 15/15°C	42.213
Ideal Gas Net Calorific Value by volume, MJ/m <sup>3</sup> at 15/15°C	38.154
Real Gas Gross Calorific Value by volume, MJ/m <sup>3</sup> at 15/15°C	42.419
Real Gas Net Calorific Value by volume, MJ/m <sup>3</sup> at 15/15°C	38.342
Ideal Gas Gross Calorific Value by molar, kJ/mol at 15°C	998.108
Ideal Gas Net Calorific Value by molar, kJ/mol at 15°C	902.139
Real Gas Gross Calorific Value by molar, kJ/mol at 15°C	998.108
Real Gas Gross Calorific Value by molar, kJ/mol at 15°C	902.139
Ideal Gas Gross Calorific Value by mass, MJ/kg at 15°C	54.58
Ideal Gas Net Calorific Value by mass, MJ/kg at 15°C	49.332
Real Gas Gross Calorific Value by mass, MJ/kg at 15°C	54.58
Real Gas Net Calorific Value by mass, MJ/kg at 15°C	49.332
Ideal Gas Wobbe Index, MJ/m <sup>3</sup> , at 15°C	53.128
Real Gas Wobbe Index, MJ/m <sup>3</sup> , at 15°C	53.212
Pressure, bar a	
Temperature, °C	-159.2
Density, kg/m <sup>3</sup> , at -159.2°C by COSTALD method	459.082
Density of real gas, kg/m <sup>3</sup> at 15°C, 101.325 kPa	0.77549